



The Carbon Footprint of Carbon Dioxide

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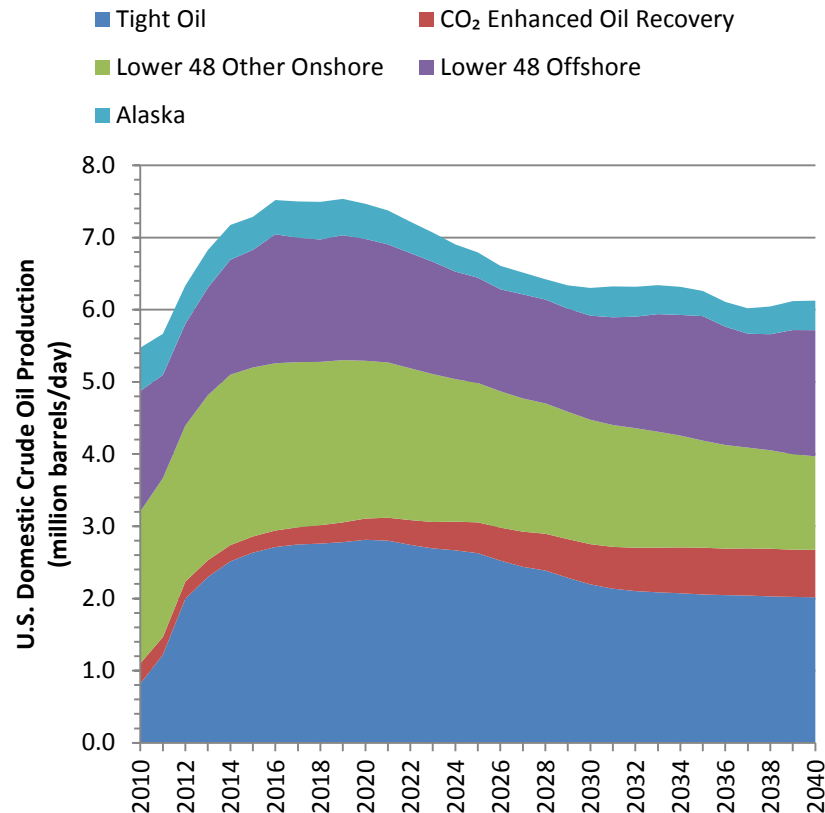
Summary



Background – Why are we studying this?

Enhanced Oil Recovery

- Enables additional recovery of crude
- CO₂ injection is alternated with injection of brine
- EOR produces additional crude; however, EOR also sequesters CO₂
- Increasing EOR production increases the demand for CO₂
- This presentation focuses on the options for acquiring CO₂ for EOR, not EOR itself (see other NETL talks)



- EOR
 - Projected annual growth rate of 3.5% through 2040
 - 10.8% of U.S. production by 2040 (compared to 5.1 % in 2010)

Technology Descriptions

- **CO₂ from Natural Dome**

- Reservoirs of high purity CO₂
- Existing CO₂ domes: McElmo, Sheep Mountain, Jackson, and Bravo domes in Western U.S.
- Recovery of CO₂ requires construction of a well with a carbon steel casing
- Contains water and must be dehydrated prior to compression and pipeline transport

- **CO₂ from Natural Gas (NG) Processing**

- Unprocessed NG contains acid gas, including variable concentrations of CO₂
- NG processing increases the heating value and reduces the acid gas composition of natural gas
- Most NG processing plants vent NG, but at some scales it may be feasible to capture CO₂

- **CO₂ from Ammonia Production**

- CO₂ is a co-product of synthetic ammonia
- Ammonia plants use natural gas as a fuel and feedstock
- An ammonia plant has two key sources of CO₂
 - Emissions from *reforming*
 - Emissions from *stripping*.
- CO₂ from reforming cannot be easily captured, but acid gas from stripping is 99 percent CO₂ and can be easily captured.

- **CO₂ from Captured Electricity Production**

- CO₂ is a co-product of fossil power generation; consider Supercritical Pulverized Coal (SCPC), and Natural Gas Combined Cycle (NGCC) power plants

Unit Processes for CO₂ from Natural Dome

- **CO₂ well construction**

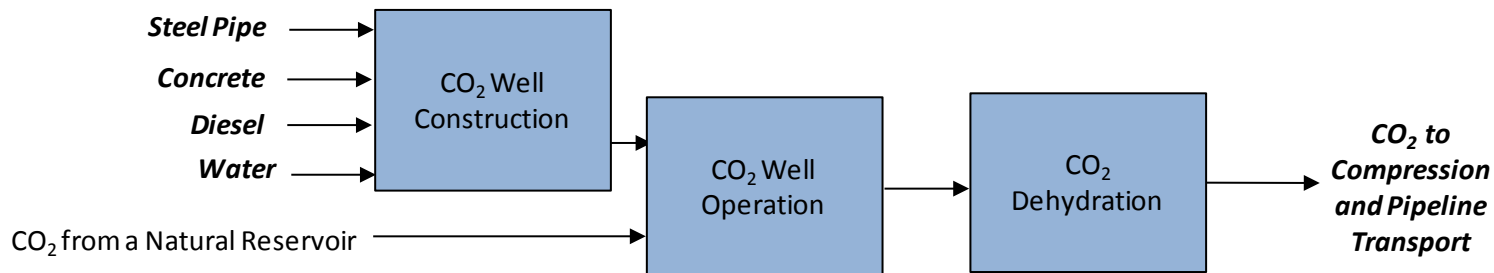
- Based on environmental impact statement for Kinder Morgan CO₂ extraction sites in Western U.S.
- Key parameters include well depth, well life, and well production rate
- Inputs include construction materials (steel and concrete), diesel used by drilling rig, and water used for drilling mud

- **CO₂ well operation accounts for fugitive CO₂ emissions**

- Valve leakage and other fugitive CO₂ emissions are accounted for by single emission factor, adapted from NETL's existing unit processes for NG extraction
- Existing NG emission factor was adapted according to molecular weights of methane vs. CO₂

- **CO₂ dehydration**

- Reboiler heat and pump power provided by grid electricity instead of NG



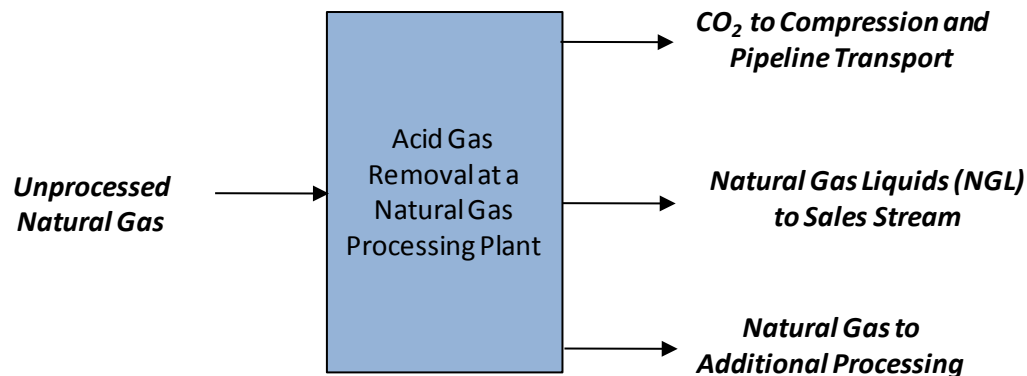
Key Parameters for CO₂ from Natural Domes

Parameter Name	Low	Expected	High	Units	Description
CO ₂ Well Construction					
Drill speed	1.42E+01	1.78E+01	2.13E+01	m/h	Drilling rate
Drill depth	1.00E+03	2.08E+03	2.50E+03	m	Well depth
Drill power	4.47E-01			MW	Power of drilling equipment in brake specific power
Diesel rate	2.21E+02			kg/MWh	Use rate of diesel; kg of diesel combusted per MWh of brake drilling energy
Total casing mass	1.03E+05			kg/well	Total mass of carbon steel well casing
Total concrete mass	1.11E+05			kg/well	Total mass of concrete well casing
Groundwater proportion	5.00E-01			dimensionless	Fraction of groundwater used during drilling
Surface water proportion	5.00E-01			dimensionless	Fraction of surface water used during drilling
Fresh water mass	6.65E+05			kg/well	Fresh water demand for drilling
Brine water mass	3.11E+05			kg/well	Brine water demand for drilling
CO ₂ Well Operation					
Fugitive CO ₂	4.64E-06			kg/kg	Fugitive loss of CO ₂ from valves, per kg of CO ₂ extracted
Well life	20	25	30	years	Production life of a CO ₂ well, used to calculate share of well construction per unit of CO ₂ dehydrated
CO ₂ production rate	5.66E+05	8.09E+05	1.05E+06	kg/well-day	Production rate of a CO ₂ well, used to calculate share of well construction per unit of CO ₂ dehydrated
Well success rate	0.65	0.70	0.85	dimensionless	Fraction of wells drilled that have economically viable production rates, used to calculate share of well construction per unit of CO ₂ dehydrated
CO ₂ Dehydration					
CO ₂ loss	1.15E-04			kg/kg CO ₂	CO ₂ emissions released to air during glycol regeneration, in terms of CO ₂ treated
Dehydration Power	1.93E-04			kWh/kg CO ₂	Electricity requirements for pumping and heating glycol used for dehydration, in terms of CO ₂ treated

- Well construction and operation parameters based on discussions with representatives of Kinder Morgan and comparisons between NG and CO₂ well practices
- Dehydration parameters based on comparisons between NG and natural CO₂ compositions

Unit Process for CO₂ from NG Processing

- **Based on acid gas removal (from NETL's NG model)**
 - Unlike existing NG model, CO₂ is captured instead of vented
 - Processed NG is sent to additional processing steps that are not necessary for CO₂
- **Parameters are used to account for variable CO₂ concentrations**
 - Production gas contains 1.5 to 80+ percent (by mass) CO₂
 - There are four existing sites that capture CO₂ from NG processing for purposes of EOR
 - This model uses the composition of gas from those facilities (76.9 to 81.1 percent) as the basis of the analysis, since low compositions cannot be economically captured
 - Reference flow of unit process is 1 kg of captured CO₂, so energy and material flows scale according to incoming CO₂ concentration



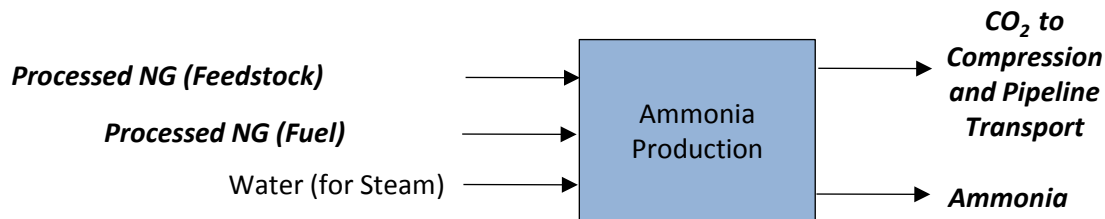
Key Parameters for CO₂ from NG Processing

Parameter Name	Low	Expected	High	Units	Description
Solvent makeup rate	9.98E-05	1.00E-04	1.01E-04	kg/kg CO ₂ captured	Makeup rate of amine solvent for CO ₂ recovery, in kg of solvent per kg of CO ₂ captured
NG fuel	6.33E-02	6.64E-02	6.95E-02	kg/kg CO ₂ captured	Combusted NG input for steam generation per unit of CO ₂ captured
Water input	1.48E-02	1.49E-02	1.50E-02	kg/kg CO ₂ captured	Water withdrawal per unit of CO ₂ captured
Surface water share	0.00E+00	5.00E-01	1.00E+00	dimensionless	Share of water withdrawn from surface water sources
CO ₂ input composition	0.8113	0.7882	0.7690	dimensionless	CO ₂ fraction of incoming stream
H ₂ S input composition	5.00E-03			dimensionless	H ₂ S fraction of incoming stream
NGL input composition	1.50E-01			dimensionless	NG liquids (NGL) fraction of incoming stream
CO ₂ pipeline composition	4.70E-03			dimensionless	CO ₂ fraction of pipeline NG, used to calculate amount of CO ₂ removed during processing
H ₂ S removal rate	9.80E-01			dimensionless	Removal rate of H ₂ S

- Solvent makeup and NG fuel rates based on variability shown by data sources (FLUOR, 2003; NETL, 2010; NETL, 2011)
- CO₂ composition in incoming gas (i.e., “production gas”) based on characteristics of NG wells that capture CO₂ for use in EOR in the Permian Basin
- CO₂ removal rate is a dependent variable, calculated based production gas composition (variable) and pipeline gas composition (0.47% mass CO₂) (NETL, 2012)

Unit Process for CO₂ from Ammonia Production

- **NG is feedstock and fuel (coal is a negligible share of ammonia feedstock in the U.S.)**
- **Ammonia production is a two-step process**
 - Step 1: Steam reforming of NG to produce carbon monoxide (CO) and hydrogen (H₂)
 - Step 2: Catalyzed conversion of hydrogen and nitrogen to ammonia
- **Instead of being used for urea production, CO₂ is sent to carbon capture, utilization and storage (CCUS)**
- **Key data sources**
 - Energy and feedstock profiles by government-sponsored research (Energetics, 2000; USDA, 2007; Worrell et al., 2000)
 - EPA emission factors for ammonia plants (EPA, 2009)
 - Water use data from European fertilizer industry (EFMA, 2000)



Key Parameters for CO₂ from Ammonia Production

Parameter Name	Low	Expected	High	Units	Description
NG input	7.78E-01	9.30E-01	1.08E+00	kg/kg CO ₂ captured	NG input (feedstock and fuel) per unit of CO ₂ captured
Water input	1.10	1.72	2.35	kg/kg CO ₂ captured	Water input per unit of CO ₂ captured
Fuel fraction	3.79E-01	4.21E-01	4.64E-01	dimensionless	Fraction of NG input used for fuel instead of feedstock

- **Total NG input is variable**

- Reformer efficiency affects amount of NG required for synthesis gas production
- Intermediate reactions that shift CO to CO₂ also affect amount of NG feedstock
- Extent of heat exchange between ammonia and urea production affects amount of NG required for fuel

- **Water input is also variable**

- Majority of water input is consumed for steam generation
- Steam requirements depend on reformer efficiency

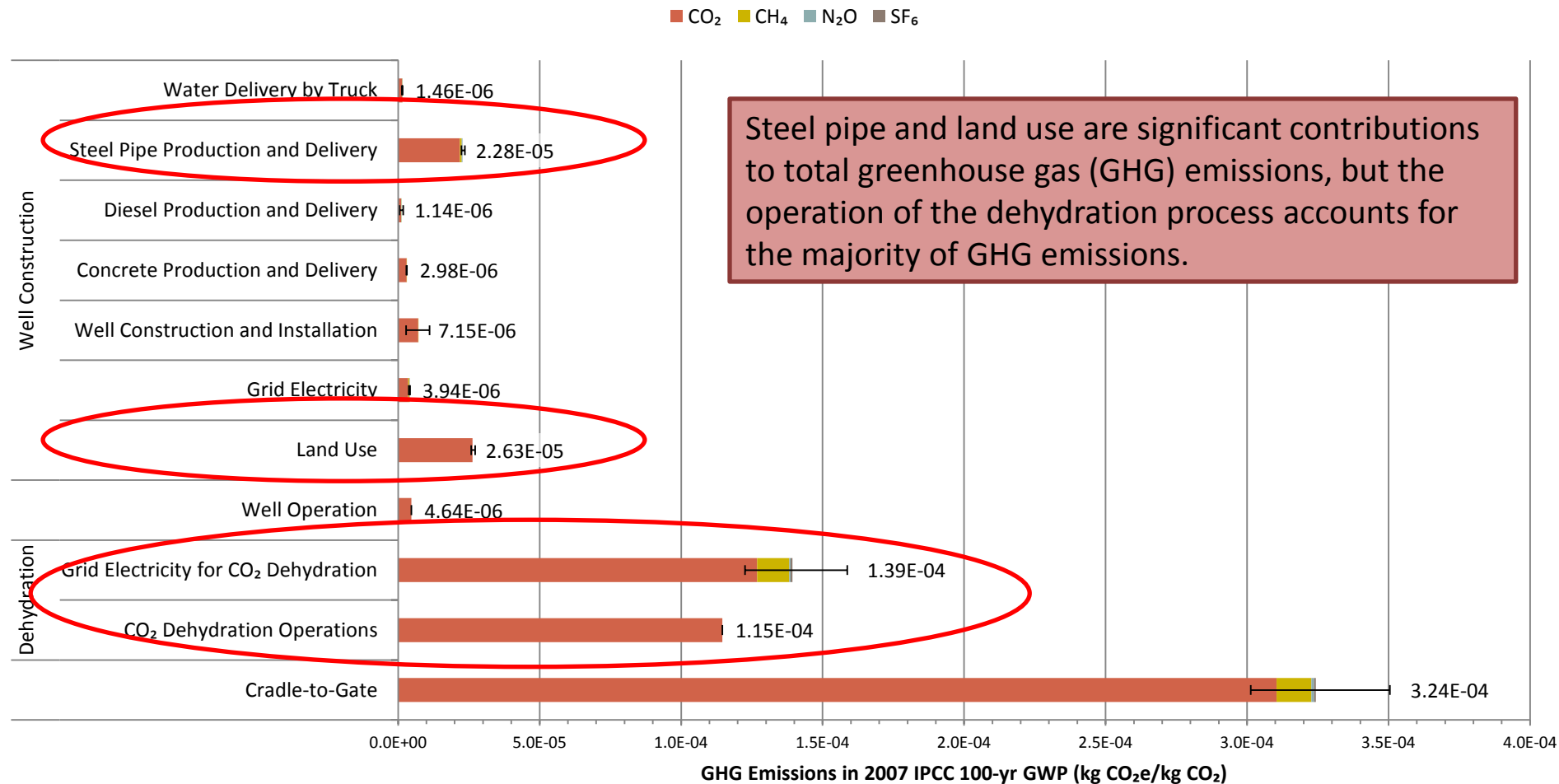
- **CO₂ production rate is also variable, but is accounted for in the NG and water input parameters**

- **Data limitations prevent parameterization of flows within ammonia plant**

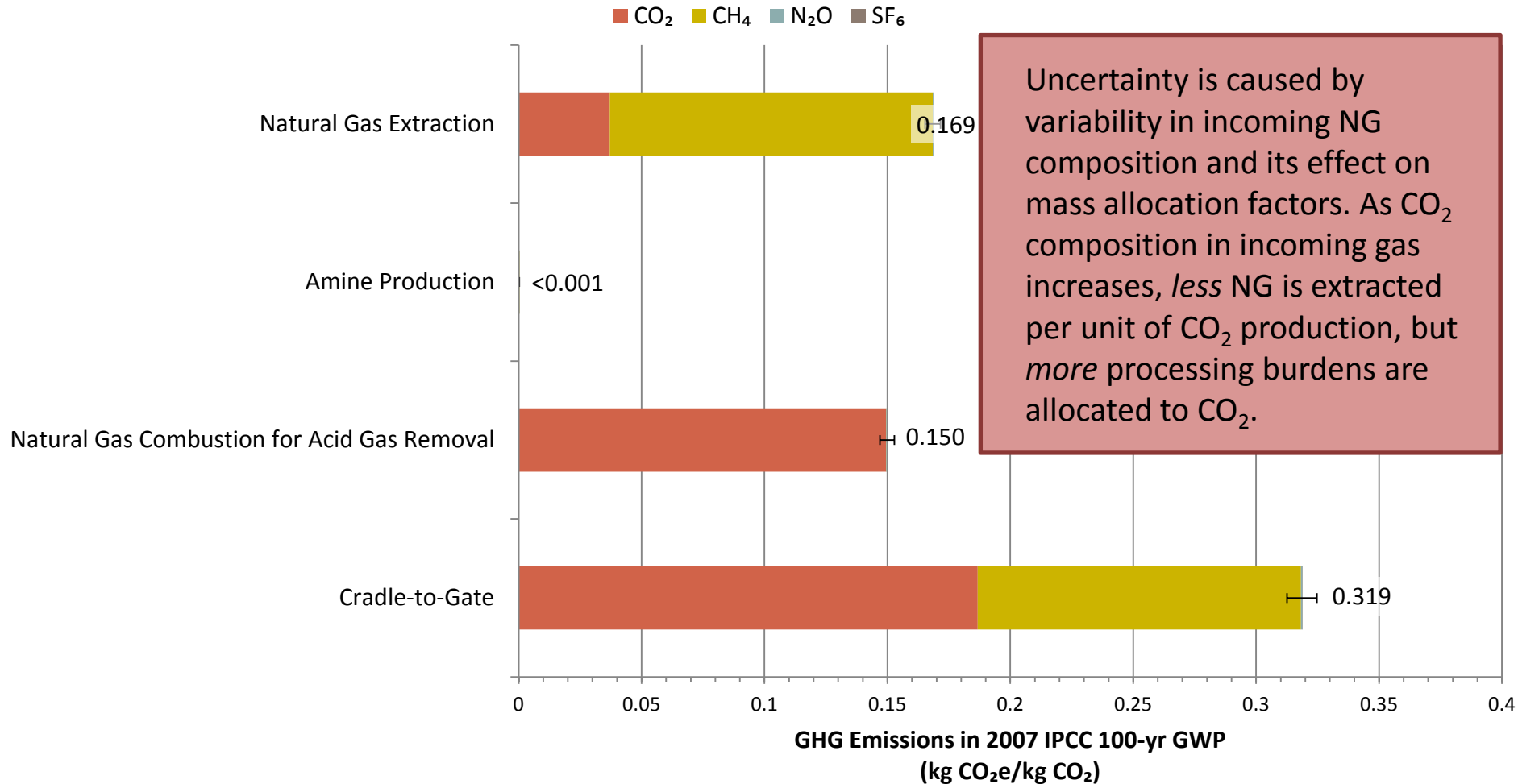
Co-Product Management

- **Natural CO₂ dome produces only CO₂ (no co-products)**
- **NG processing produces CO₂, NG, and NGL**
 - CO₂ cannot be expressed in terms of energy, so energy-based co-product allocation is *not* feasible
 - Mass-based co-product allocation is feasible and is based on masses of produced CO₂ and NG
 - System expansion is also feasible, but requires consequential assumptions
- **Ammonia plant produces CO₂ and ammonia**
 - CO₂ cannot be expressed in terms of energy, so energy-based co-product allocation is *not* feasible
 - Mass-based co-product allocation is feasible and is based on masses of produced CO₂ and ammonia
 - System expansion is also feasible, but requires consequential assumptions
- **Coal and NG power plants produce CO₂ and electricity**
 - CO₂ cannot be expressed in terms of energy, so energy-based co-product allocation is *not* feasible; nor can electricity be expressed in terms of mass
 - System expansion is the only option

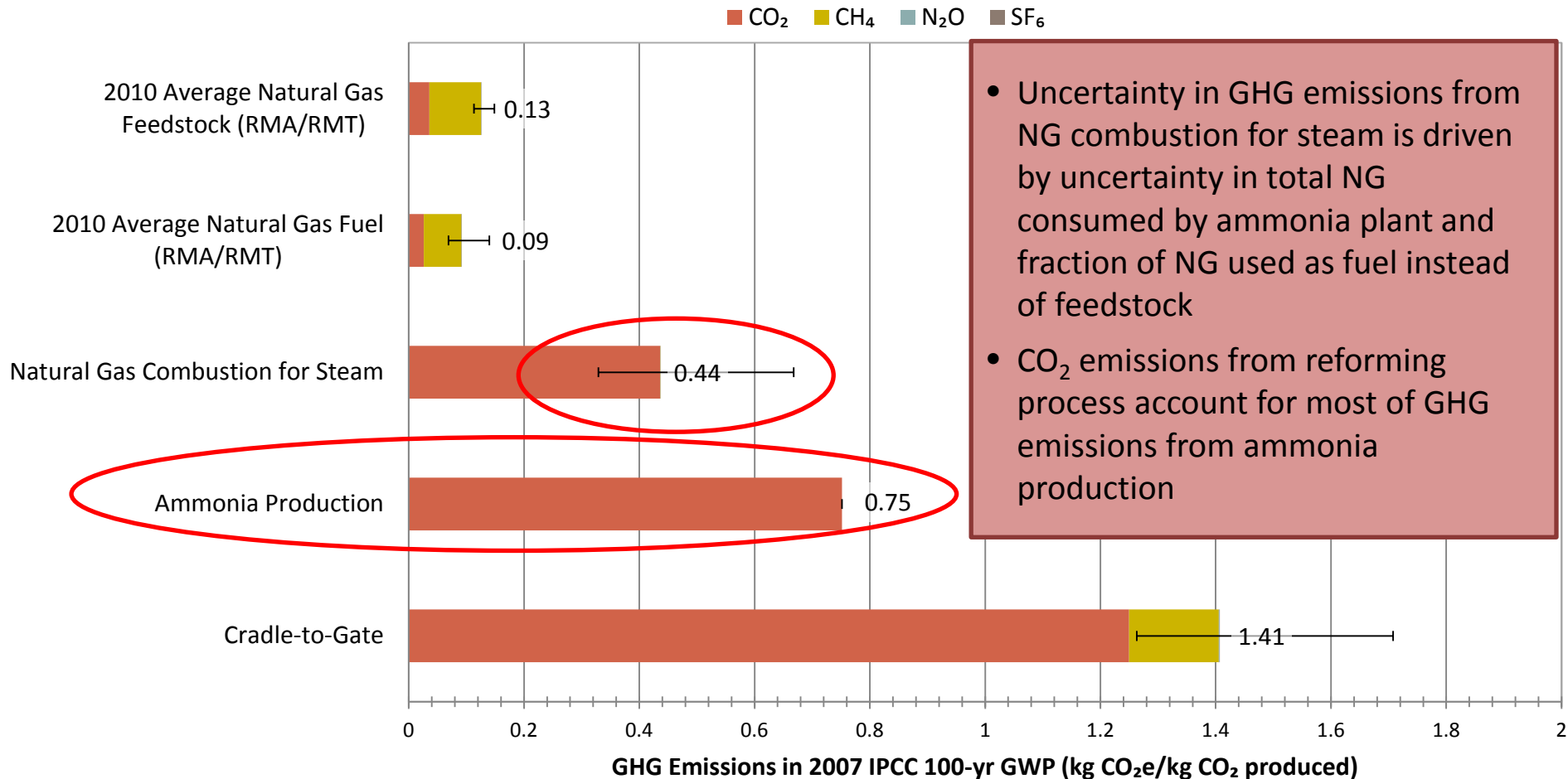
Cradle-to-Gate Results for CO₂ from Natural Dome



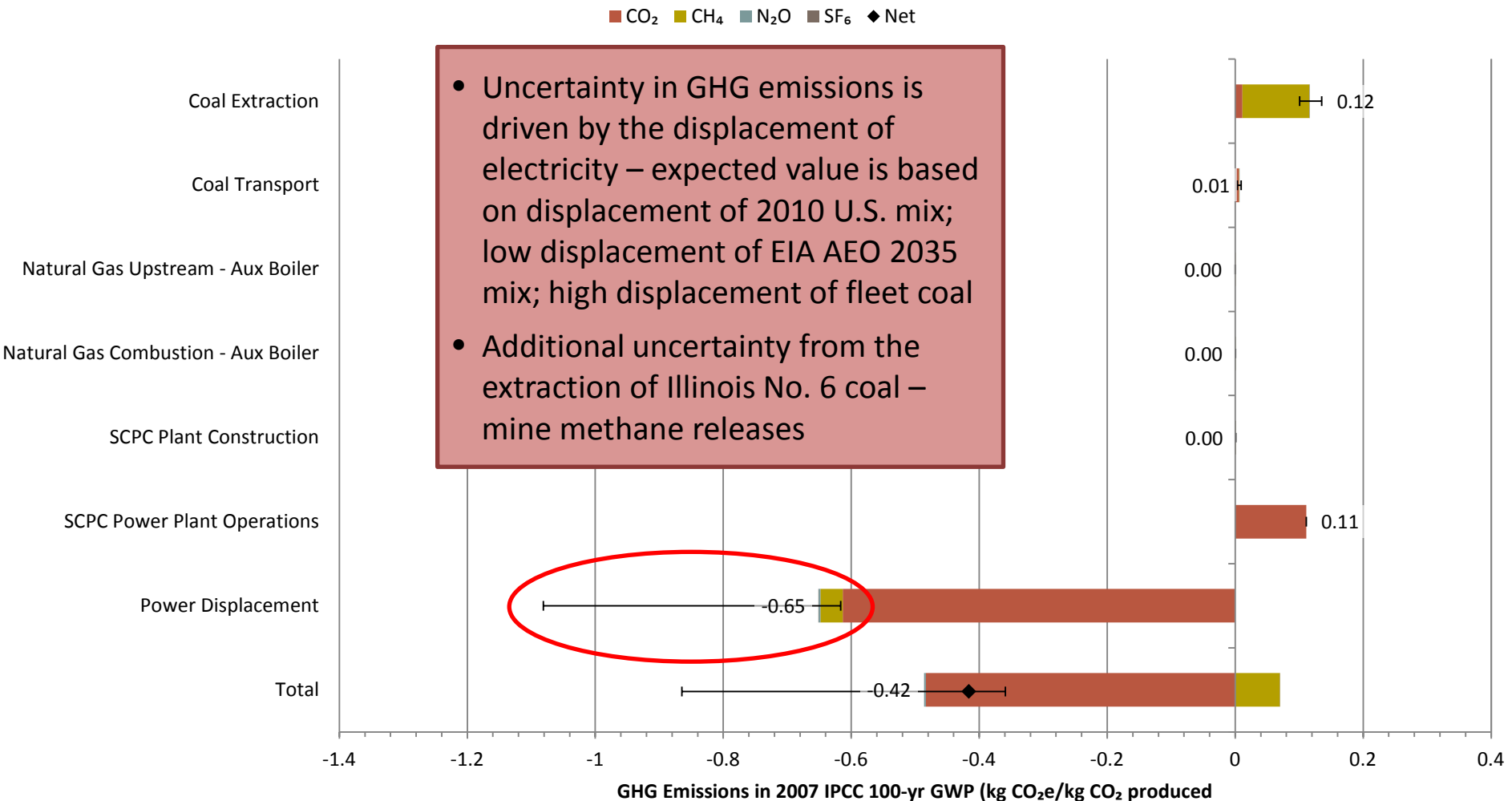
Cradle-to-Gate Results for CO₂ from NG Processing (Mass Allocation)



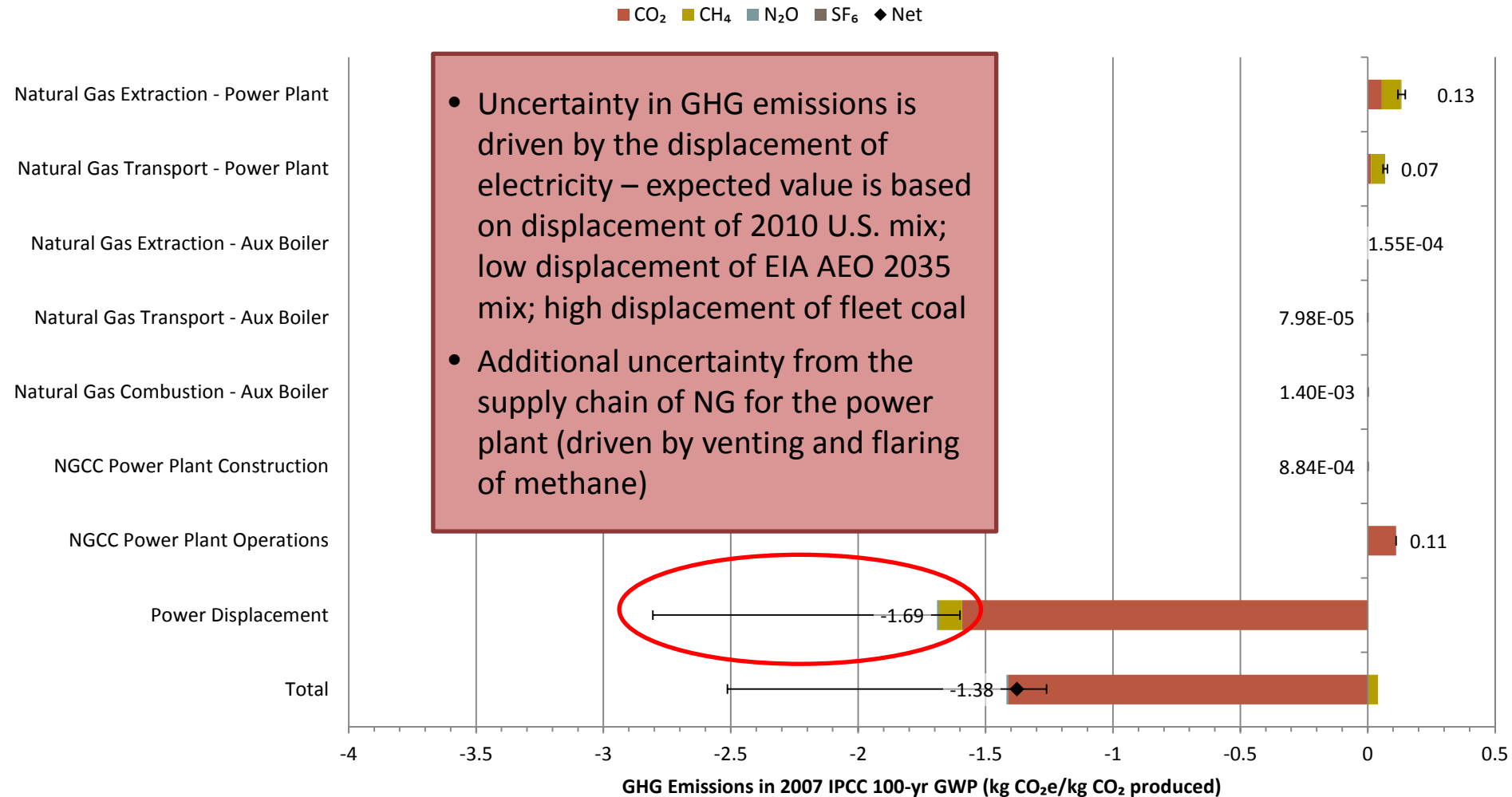
Cradle-to-Gate Results for CO₂ from Ammonia Production (Mass Allocation)



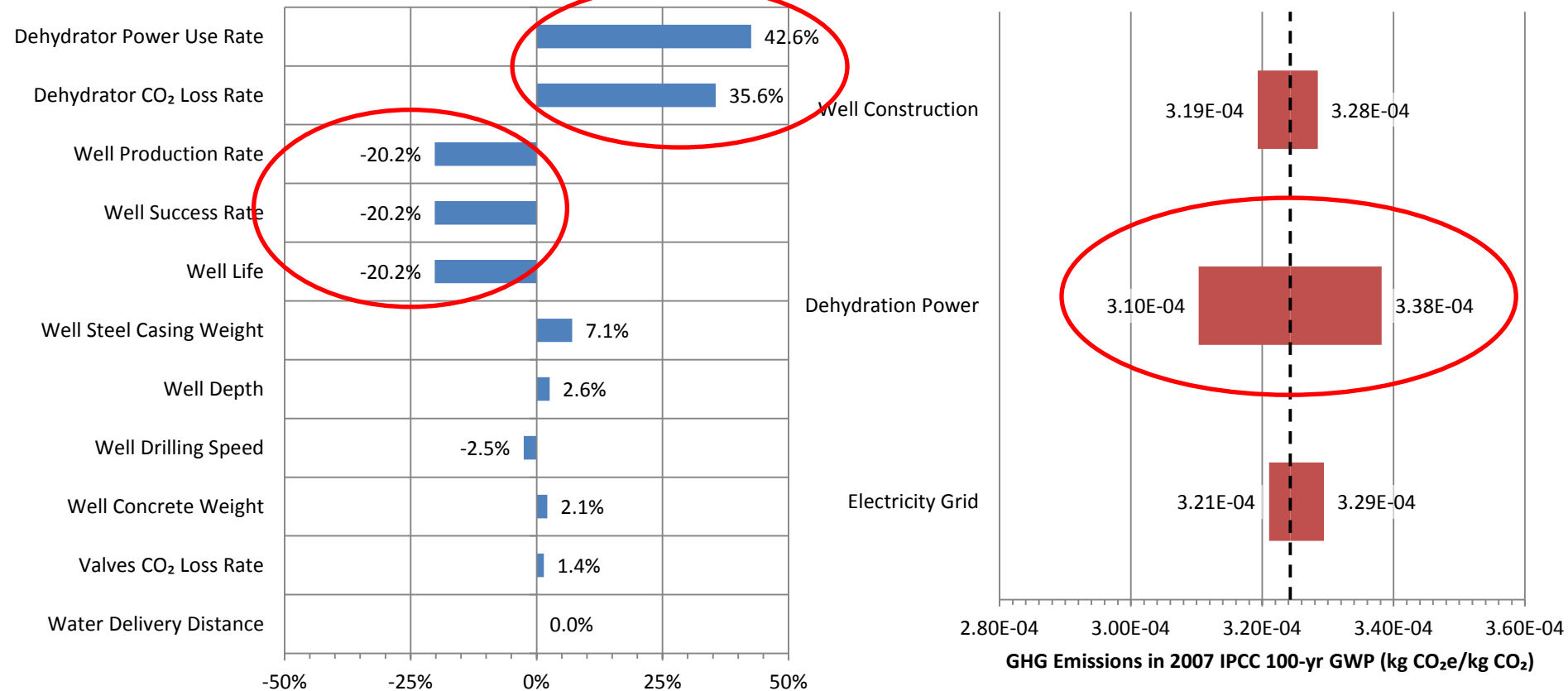
Cradle-to-Gate Results for CO₂ from SCPC Power Plant (Displacement)



Cradle-to-Gate Results for CO₂ from NGCC Power Plant (Displacement)

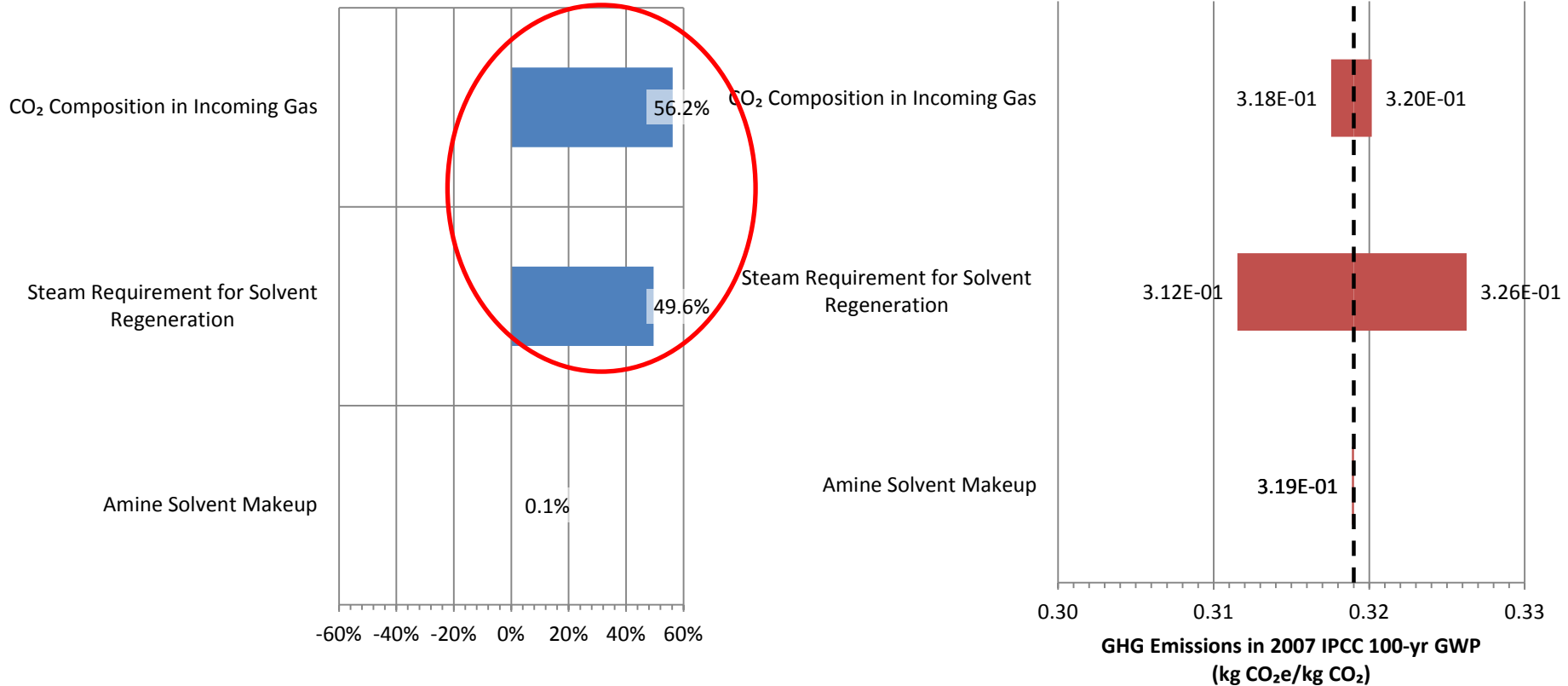


GHG Sensitivity and Uncertainty for CO₂ from a Natural Dome



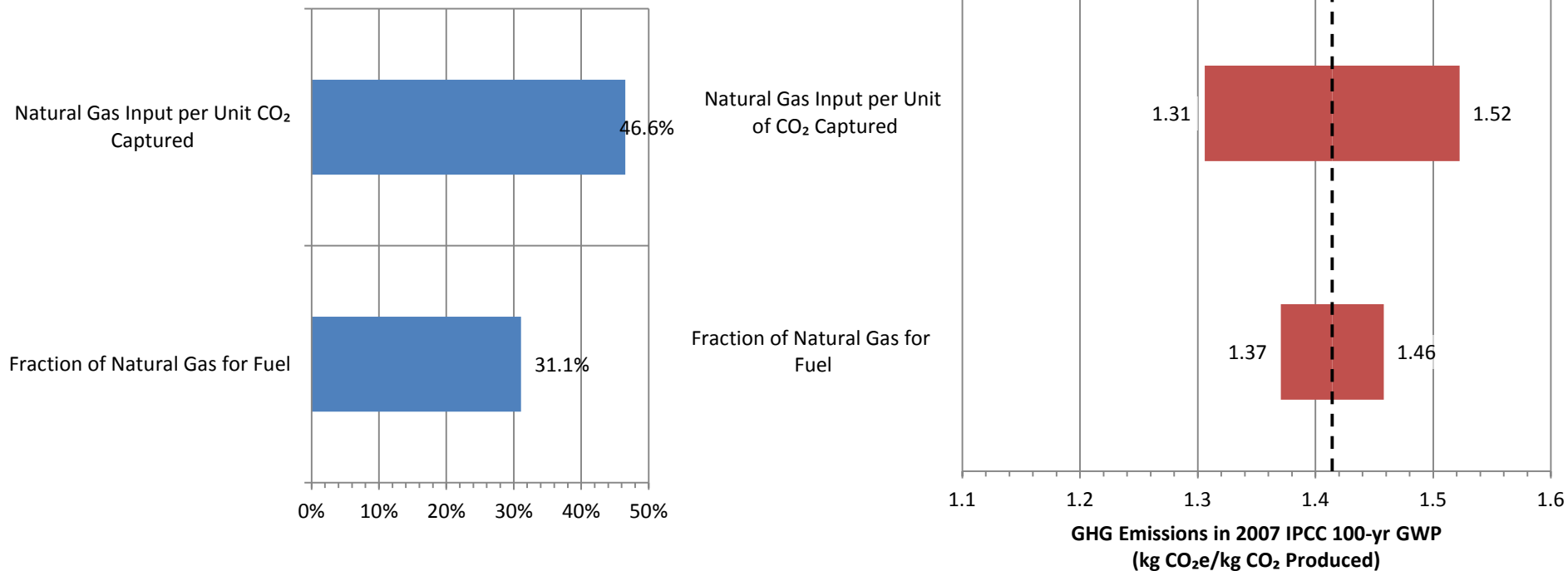
- GHG results are sensitive to changes in dehydrator variables (power use and CO₂ loss rate)
- GHG results show an inverse relationship to well production rate, well success rate, and well life – these parameters affect denominator used for apportioning construction and land use burdens
- Greatest uncertainty in GHG results is caused by uncertainty in CO₂ processing (dehydration)

GHG Sensitivity and Uncertainty for CO₂ from NG Processing



- GHG emissions sensitive to changes in CO₂ composition of incoming gas and steam rates for gas processing

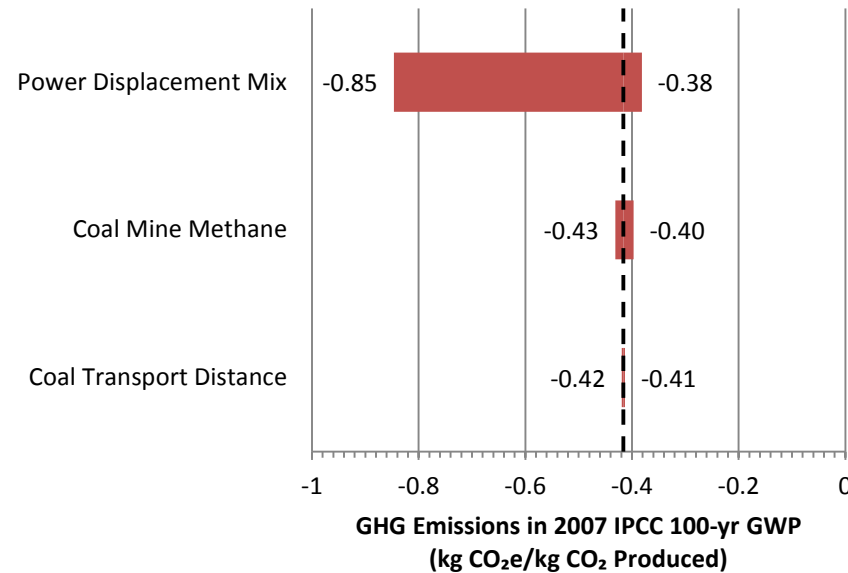
GHG Sensitivity and Uncertainty for CO₂ from Ammonia Production



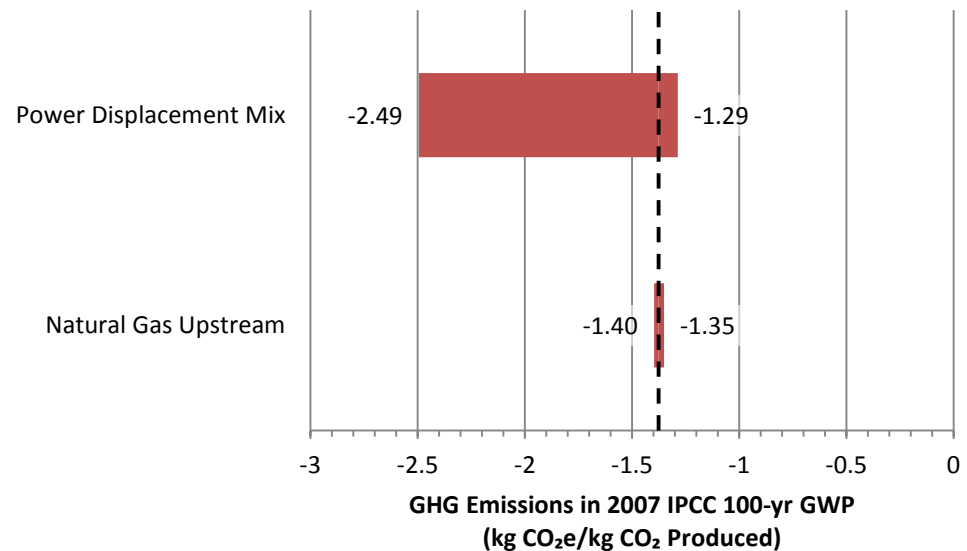
- Due to the high GHG footprint of NG extraction, GHG emission sensitivity *and* uncertainty driven by NG input rate
- Data limitations prevent parameterization of other ammonia plant operating characteristics

GHG Uncertainty for CO₂ from SCPC and NGCC Power Plants

SCPC Power Plant

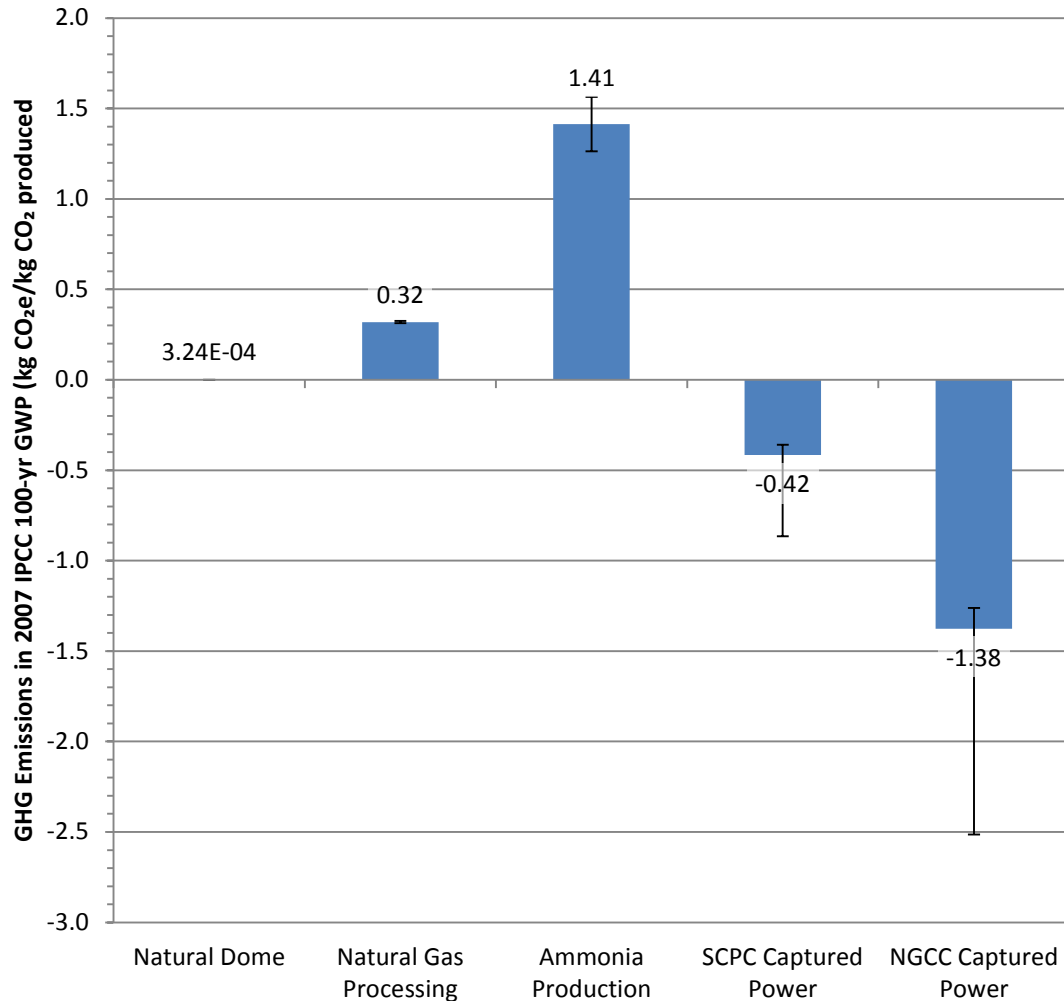


NGCC Power Plant



- For both technologies, majority of uncertainty in GHG emissions is driven by the displacement of electricity – expected value is based on displacement of 2010 U.S. mix; low displacement of EIA AEO 2035 mix; high displacement of fleet coal
- Additional uncertainty from the extraction of Illinois No. 6 coal – mine methane releases for SCPC
- Additional uncertainty from the supply chain of NG for the power plant (driven by venting and flaring of methane) for NGCC

Carbon Footprint of CO₂ Summary



- Captured power values yield negative results because of the displacement of the co-product (electricity) with a more GHG-intensive conventional source (U.S. grid mix)
- Natural Dome has effectively zero environmental burden – the resource is extracted directly, no conversion processes necessary'
- Alternative sources for captured CO₂ (NG processing and Ammonia production) are less GHG-efficient methods because of the processing emissions

Recommendations and Conclusions

- Above results are only from cradle to gate, so they should be used with care
- These models allow further LCA modeling of carbon capture, utilization, and storage (CCUS) scenarios
- A detailed report on these models is currently under review and will be released soon on the NETL Energy Analysis site
- For more information on EOR, consider attending these NETL presentations in the “Fossil Fuels 2” session from 3:00-4:30 in Ligurian II:
 - A Parameterized Life Cycle Model of Crude from CO₂-Enhanced Oil Recovery
 - The Challenge of Co-product Accounting for Large-scale Energy Systems: Power, Fuel and CO₂

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References

- EFMA. (2000). Best Available Techniques for Pollution and Control in the European Fertilizer Industry: Production of Ammonia. Brussels, Belgium.
- EIA. (2006) *Natural Gas Processing: The Crucial Link Between Natural Gas Production and Its Transportation to Market*. U.S. Energy Information Administration. Office of Oil and Gas. Washington, D.C.
- Energetics. (2000). Energy and Environmental Profile of the U.S. Chemical Industry. U.S. Department of Energy, Office of Industrial Technologies. Accessed on December 27, 2012 at http://www1.eere.energy.gov/manufacturing/resources/chemicals/pdfs/profile_chap1.pdf
- EPA. (2009) *Technical Support Document for the Ammonia Production Sector Proposed Rule for Mandatory Reporting of Greenhouse Gases*. U.S. Environmental Protection Agency. Office of Air and Radiation. Washington, D.C.
- FLUOR. (2003) FLOUR's Econamine FG Plus Technology: An Enhanced Amine-Based CO₂ Capture Process, FLUOR Corporation. Accessed on July 30, 2012 at http://www.fluor.com/SiteCollectionDocuments/FluorEconamineFGPlusTechnology-NETLConf_May2003.pdf.
- MIT. (2011). Lebarge Fact Sheet: Carbon Dioxide Capture and Storage Project. Massachusetts Institute of Technology. Retrieved December 11, 2012, from http://sequestration.mit.edu/tools/projects/la_barge.html
- NETL. (2010) Cost and Performance Baseline for Fossil Energy Plants, Volume 1: Bituminous Coal and Natural Gas to Electricity, National Energy Technology Laboratory, Pittsburgh, PA. Accessed on July 30, 2012 at http://www.netl.doe.gov/energy-analyses/pubs/BitBase_FinRep_Rev2.pdf.
- NETL. (2011) DOE/NETL Advanced Carbon Dioxide Capture R&D Program: Technology Update, National Energy Technology Laboratory, Pittsburgh, PA. Accessed on July 30, 2012 at http://www.netl.doe.gov/technologies/coalpower/ewr/pubs/CO2Handbook/CO2-Capture-Tech-Update-2011_Front-End%20Report.pdf.
- NETL. (2012). Role of Alternative Energy Sources: Natural Gas Technology Assessment. (DOE/NETL-2012/1539). Pittsburgh, PA: National Energy Technology Laboratory. Retrieved November 8, 2012, from <http://www.netl.doe.gov/energy-analyses/pubs/NGTechAssess.pdf>
- USDA. (2007). Impact of Rising Natural Gas Prices on U.S. Ammonia Supply. Retrieved November 21, 2012, from http://www.ers.usda.gov/media/198815/wrs0702_1_.pdf
- Worrell, E., Philipsen, D., DanEinstein, & Martin, N. (2000). Energy Use and Energy Intensity of the U.S. Chemical Industry: Ernest Orlando Lawrence Berkeley National Laboratory.